

# Na-ion Anode Development

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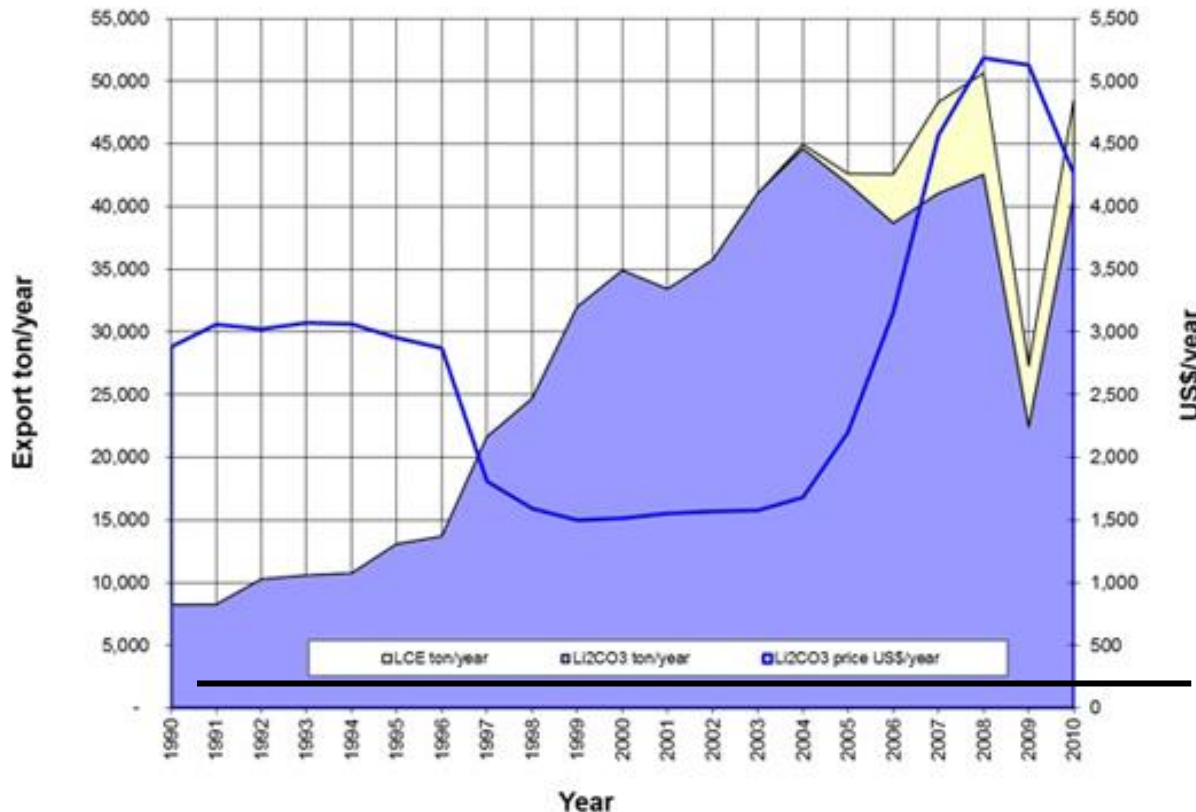
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The Pennsylvania State University  
September 17, 2014

PENNS<sup>T</sup>ATE<sup>®</sup>



Sodium-ion batteries could be cheaper than Li-ion batteries, and use a much more abundant resource.

**Li<sub>2</sub>CO<sub>3</sub> (Li precursor) price, 1990-2010**

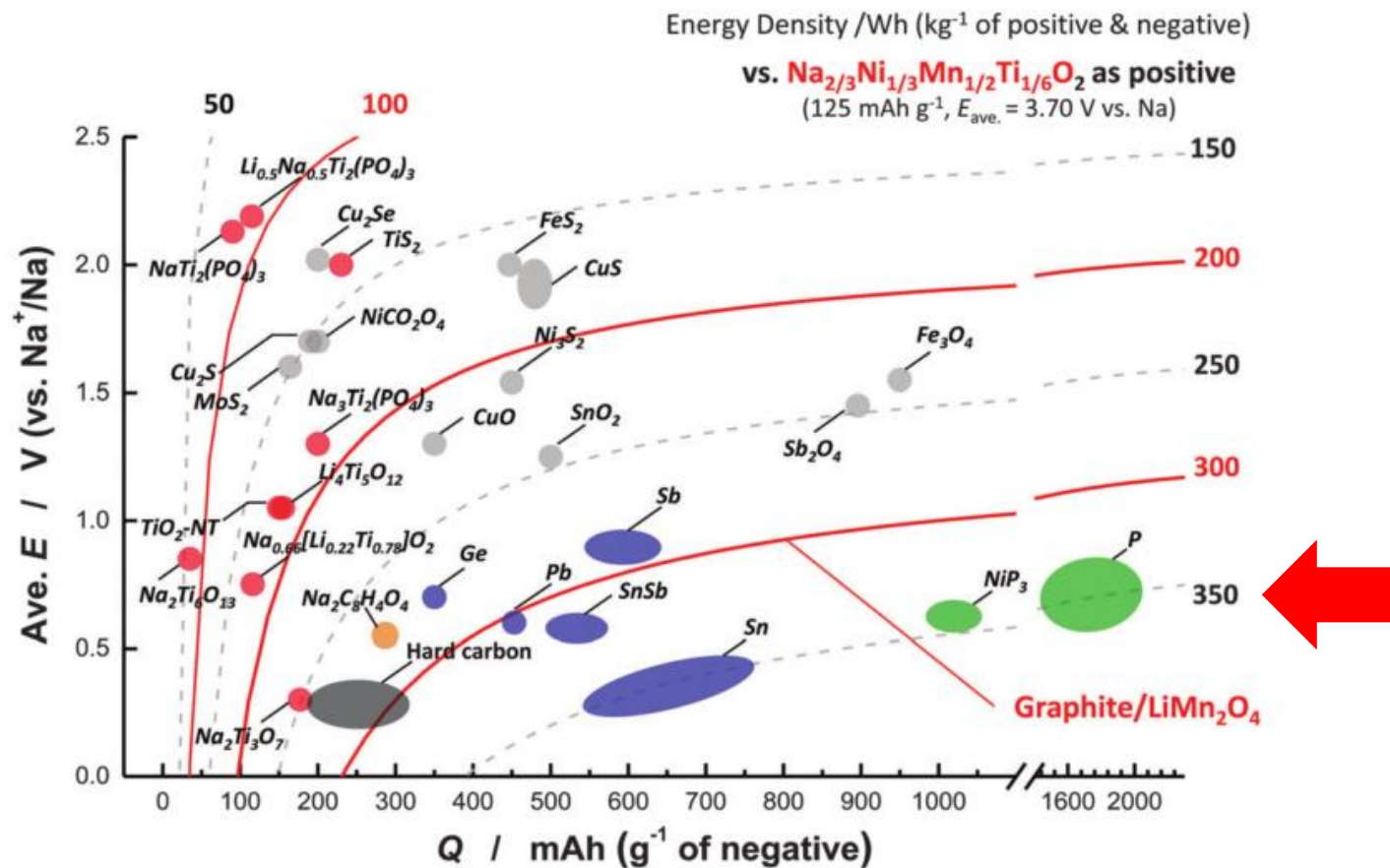


**Known global supply (tons):**  
 Li<sub>2</sub>CO<sub>3</sub>: ~135 million  
 Na<sub>2</sub>CO<sub>3</sub>: ~47 billion

**Na<sub>2</sub>CO<sub>3</sub> price today:**  
 ~\$200/ton

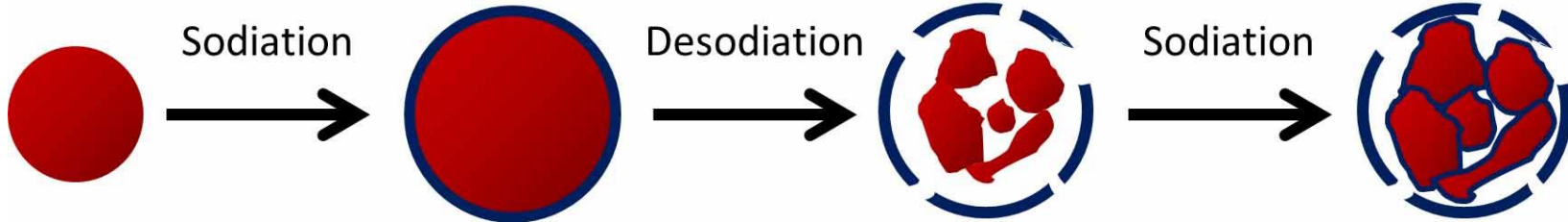
Plot from <http://www.lithiumsite.com/market.html>  
 Values from Evans, An Abundance of Lithium, 2008; Dolley, Soda Ash, U.S. Geological Survey Mineral Commodity Summaries, 2014; alibaba.com, <http://www.alibaba.com/showroom/sodium-carbonate-price-per-ton.html>, accessed August 24, 2014

Phosphorus is a promising material for sodium-ion batteries thanks to its high capacity and low cost.



# Issues

- Low electrical conductivity ( $1 \times 10^{-14}$  S/cm)
- Large volume expansion



Particle pulverization

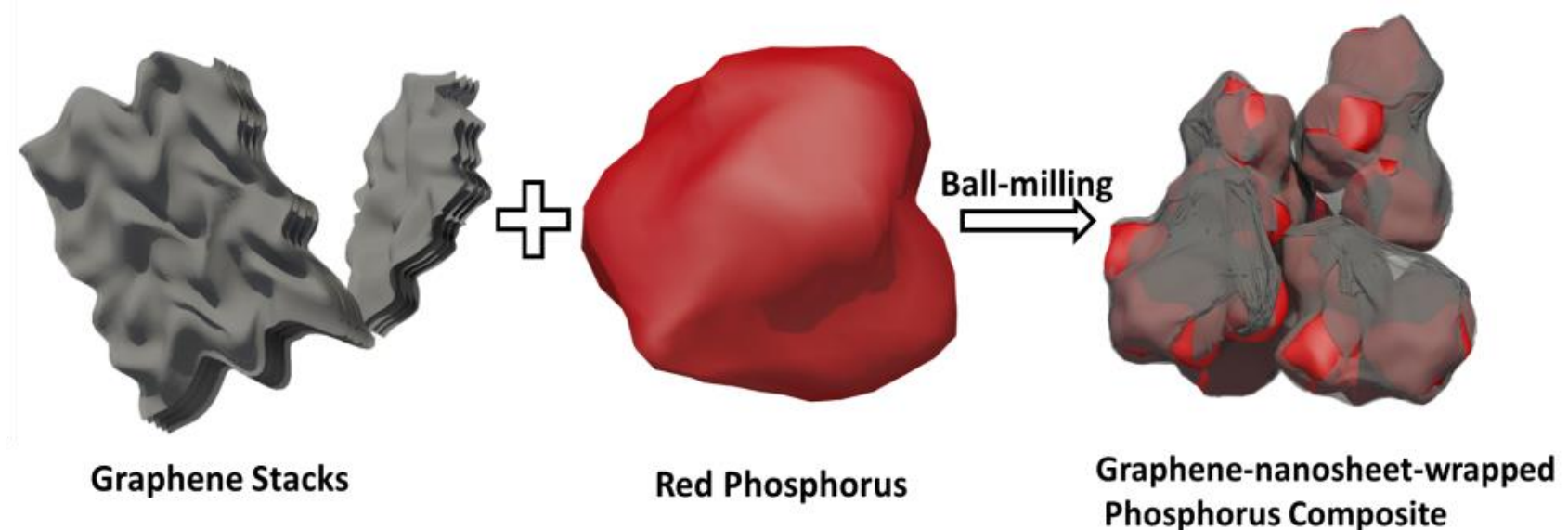
Loss of electrical contact

Unstable SEI

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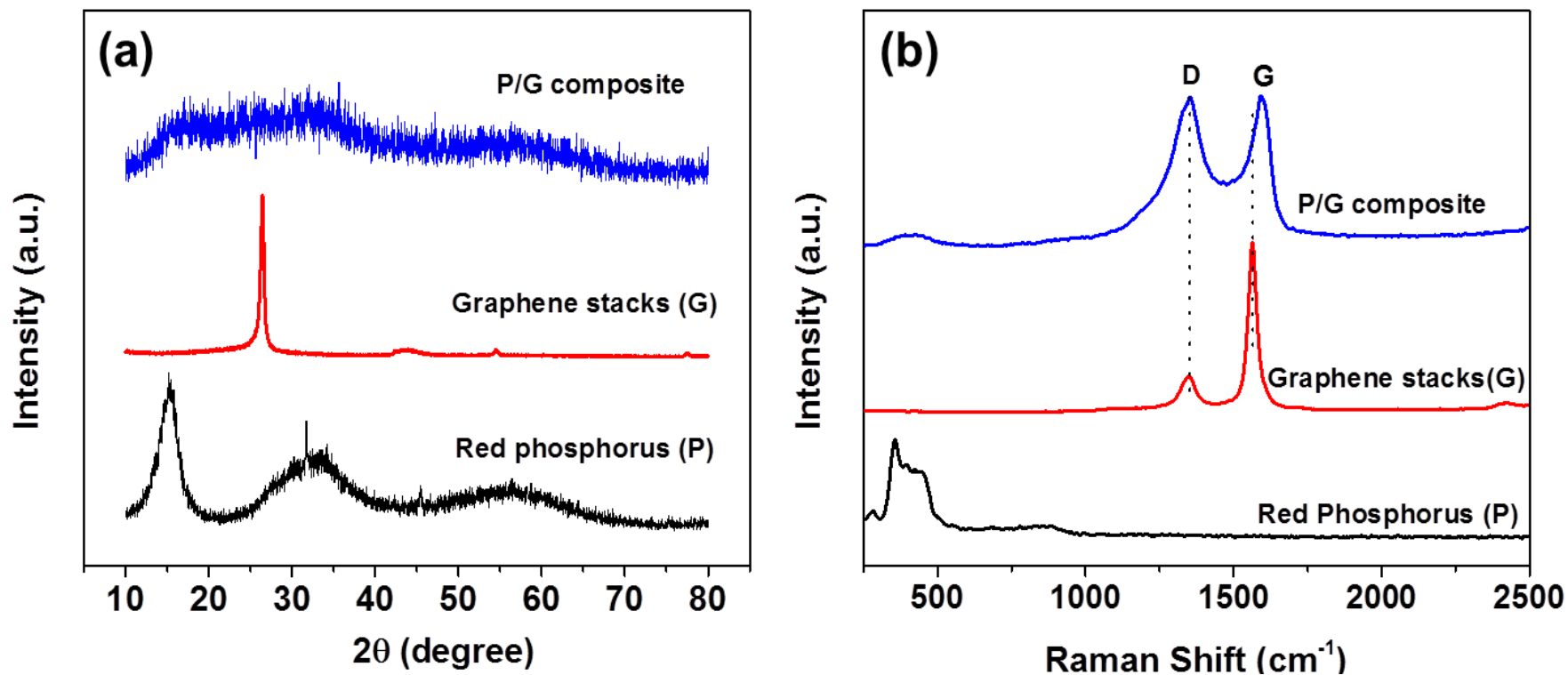
Element	Max. Na content in $\text{Na}_x\text{X}$	Vol. ratio of $\text{Na}_x\text{X}/\text{X}$
Na	–	–
C	$< \text{NaC}_6$	$< 2\%$
Si	NaSi	130%
Sn	$\text{Na}_{3.75}\text{Sn}$	525%
Sb	$\text{Na}_3\text{Sb}$	390%
Pb	$\text{Na}_{3.75}\text{Pb}$	487%
P	$\text{Na}_3\text{P}$	491%

# Graphene-nanosheet-wrapped Phosphorous Composite Anode



- (1) a facile, low-energy ball milling technique was adopted (400 rpm for 1000 min).
- (2) Low cost commercial red phosphorus and graphene stacks as raw materials.

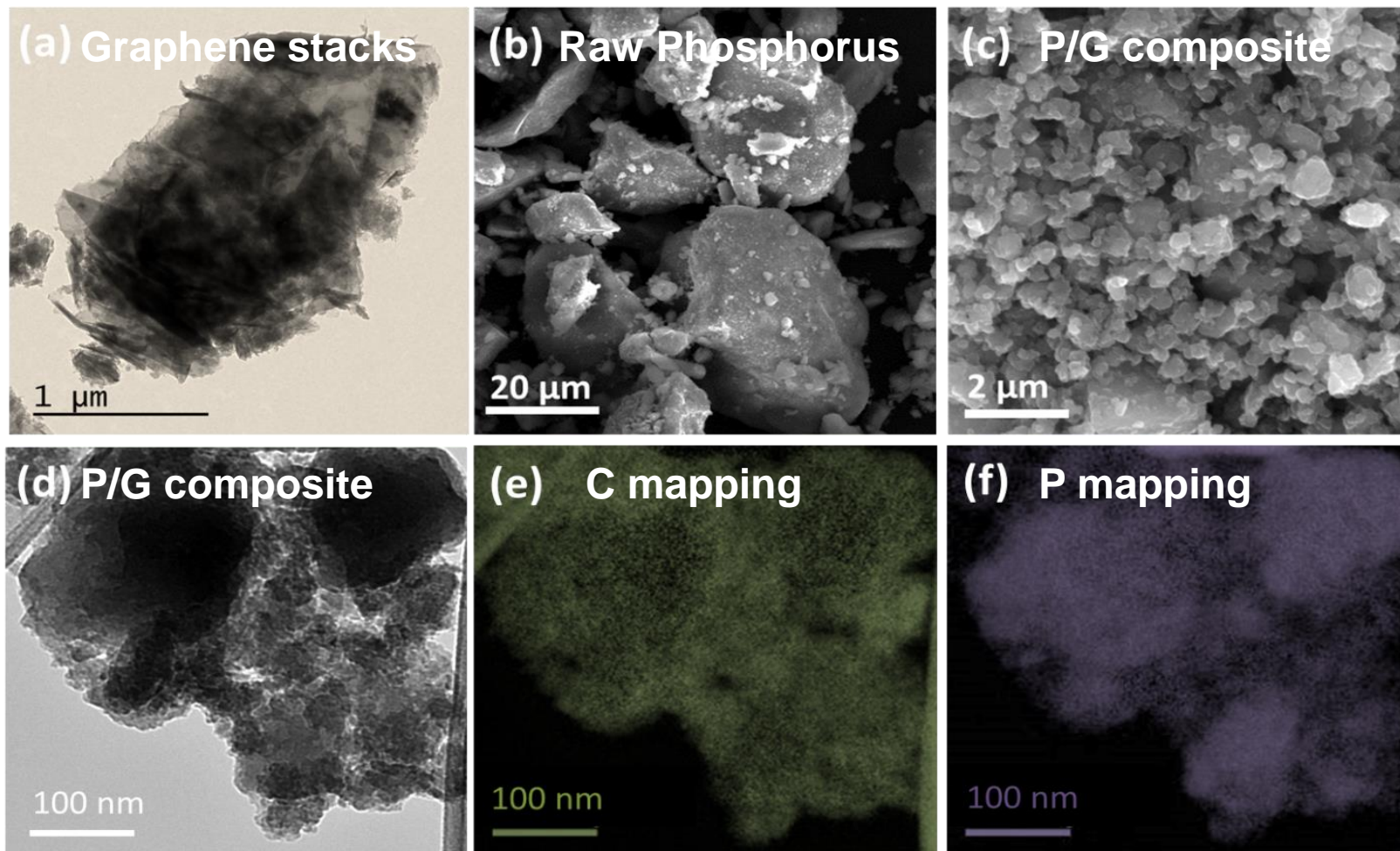
# Characterization



- (1) Amorphous structure structure of the P/G composite.
- (2) Graphenen stacks were exfoliated to graphene nanosheets driven by mechanical shearing upon ball-milling.

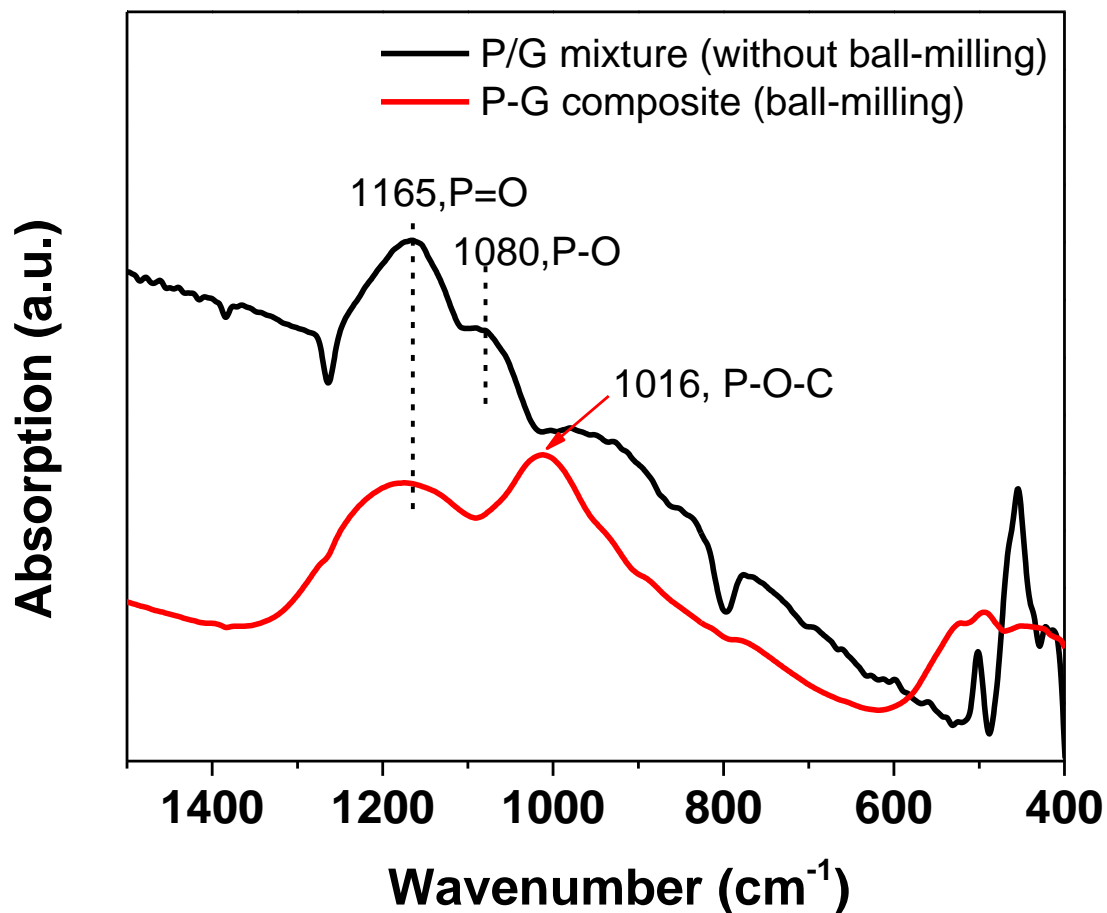


# Morphology



- (1) The graphene nanosheets can form a conductive matrix within the composite.
- (2) The large bulk phosphorus were grinded into microscale or nanoscale particles.

# Chemical Interaction between Phosphorous and Graphene Sheets

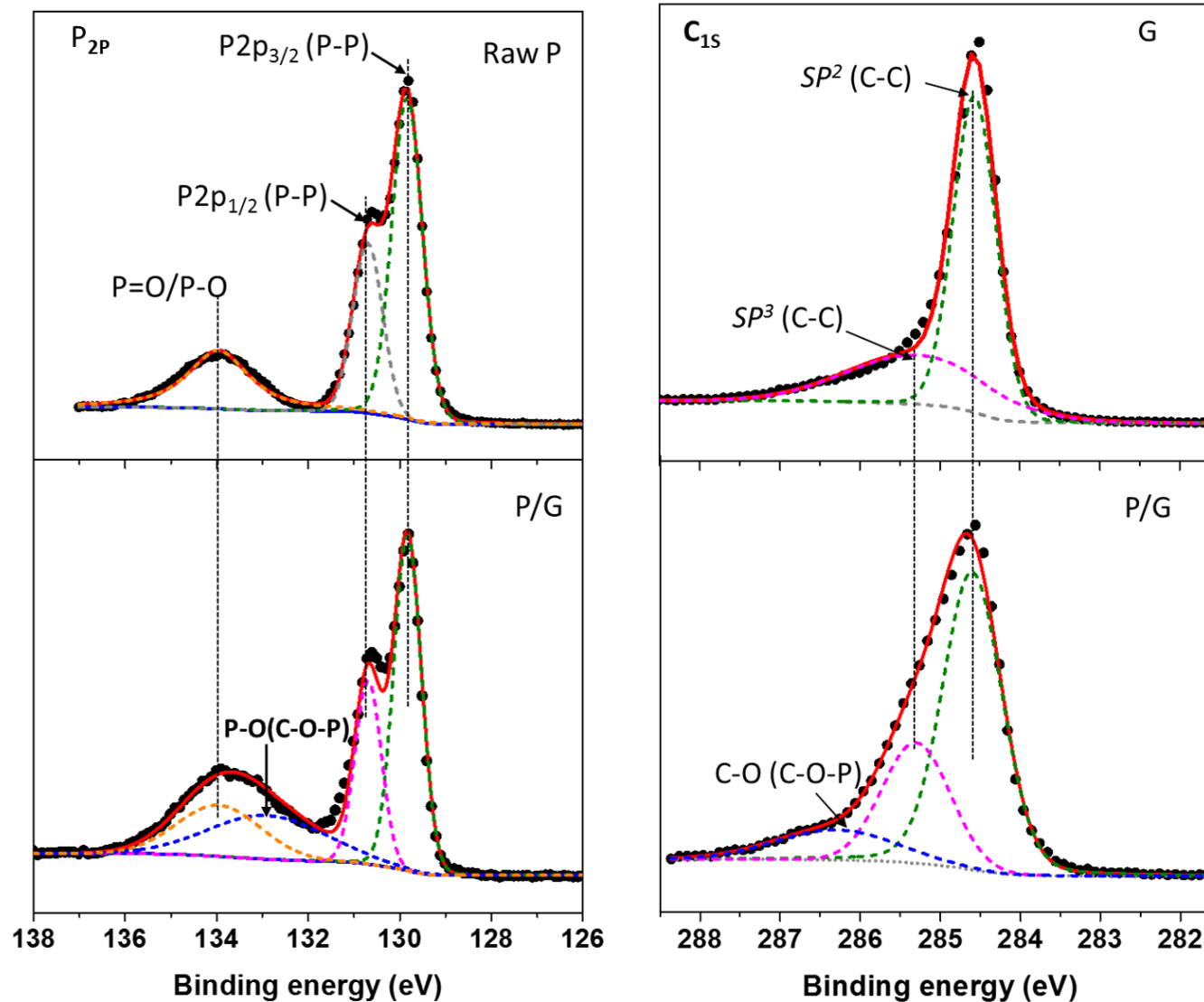


Formation of **P-O-C** bond between phosphours and graphene nanosheets via ball-milling

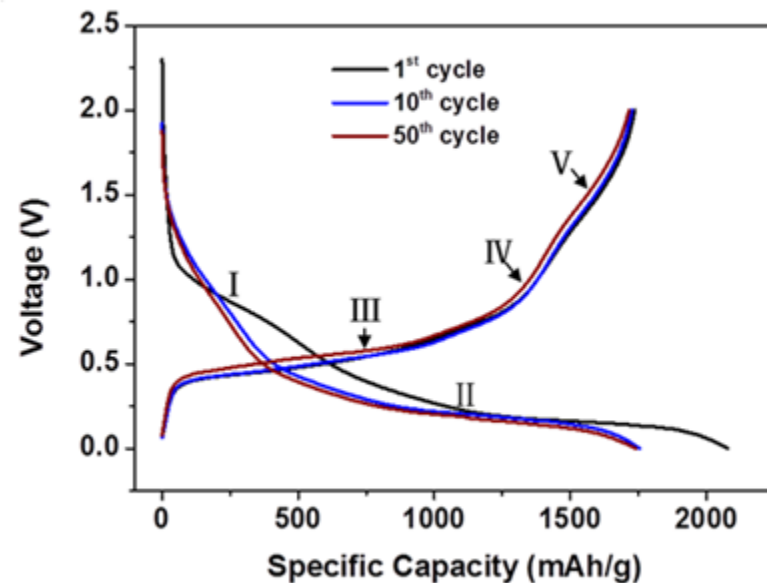
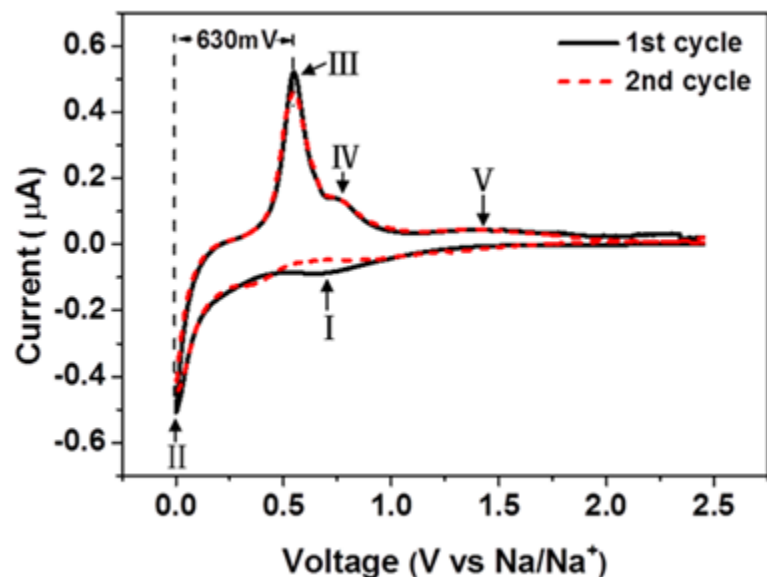
The graphene nanosheets can chemically bind with phosphorus during milling process, which facilitates the intimate contact of graphene nanosheets with phosphorus particles.



# Chemical Interaction between Phosphorous and Graphene Sheets

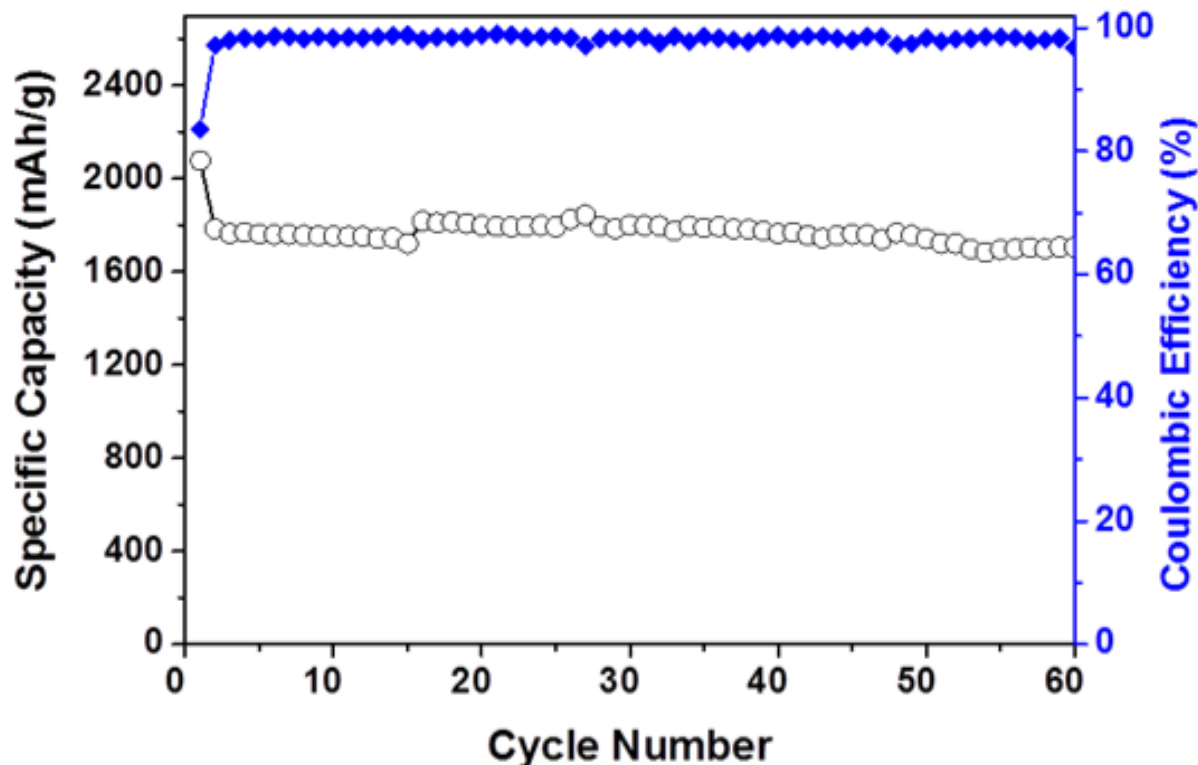


Cyclic voltammetry and voltage profiles indicate stable electrochemical behavior with cycling.



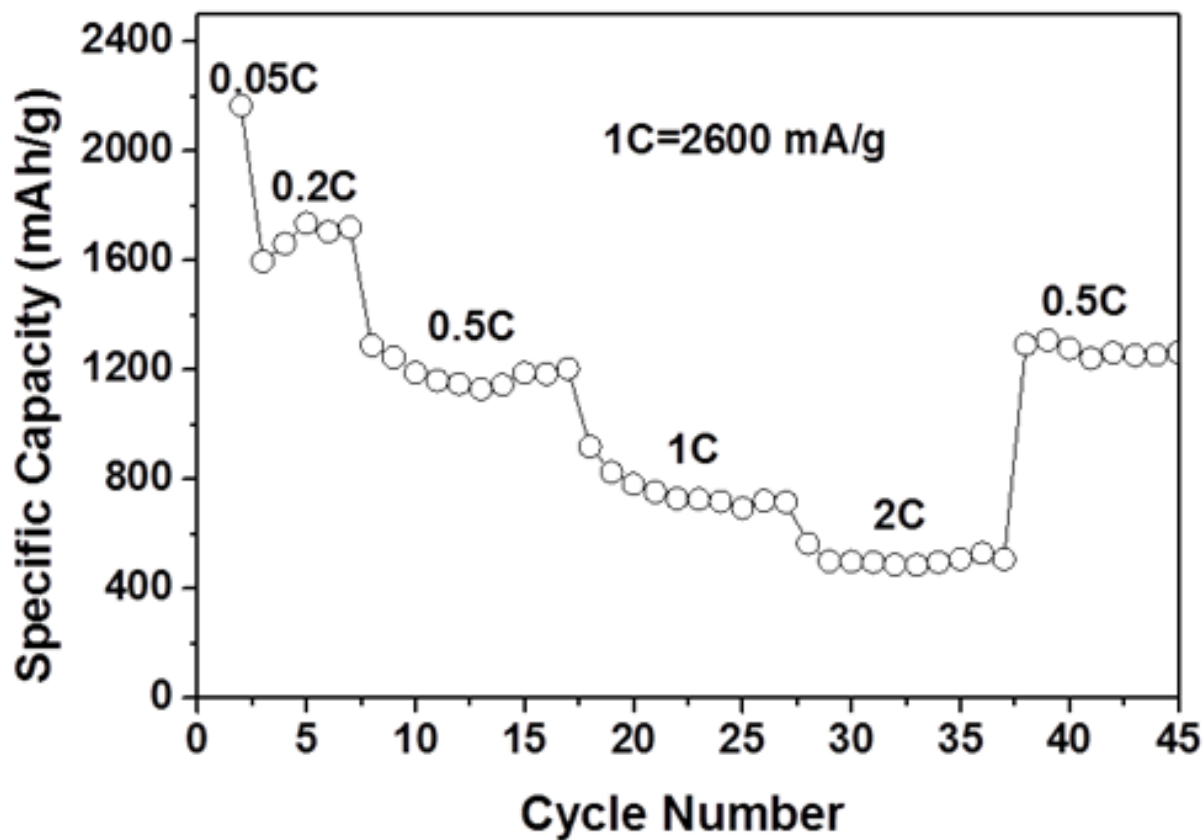
- I) SEI formation
- II) Na<sub>x</sub>P formation upon sodiation
- III-V) Na<sub>2</sub>P, NaP, and NaP<sub>7</sub> formation upon desodiation

A high capacity and coulombic efficiency are achieved using graphene-wrapped phosphorous.

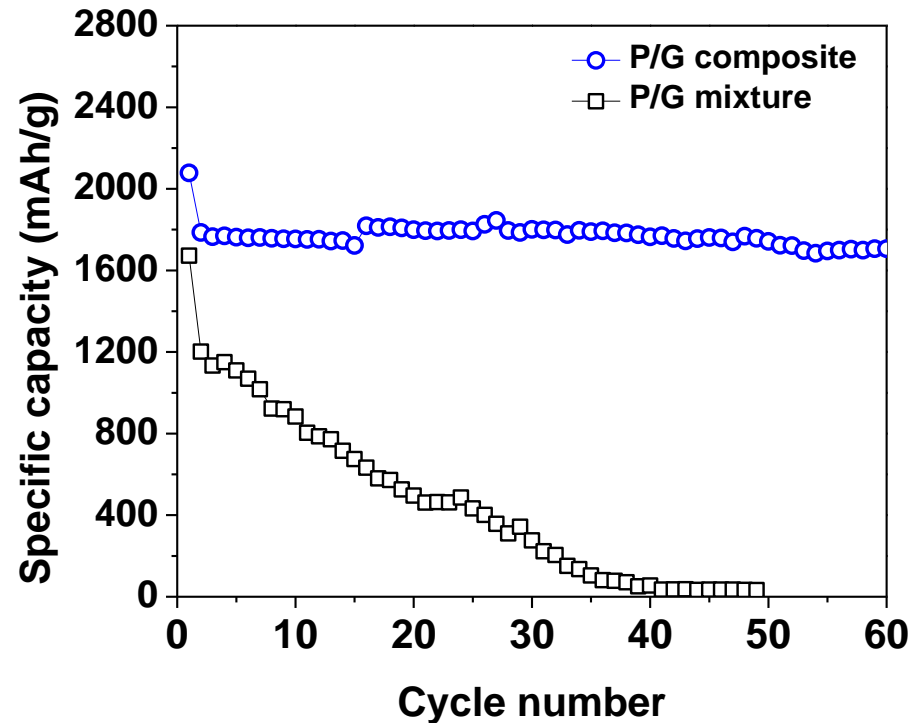
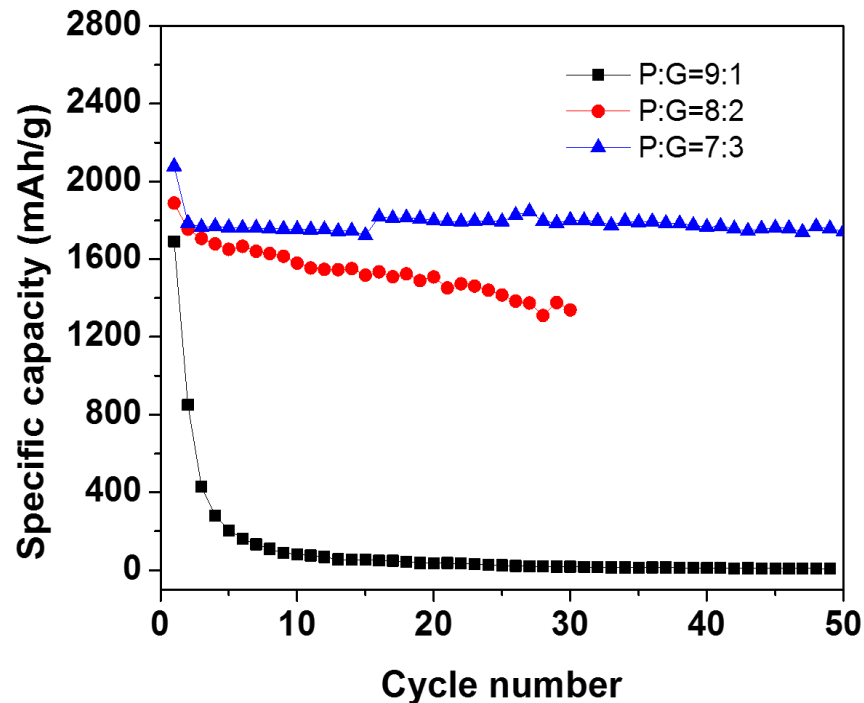


Initial Capacity	2077 mAh/g
Initial Columbic Efficiency	83%
60 <sup>th</sup> Cycle Capacity	1706 mAh/g

The rate performance for graphene-wrapped phosphorous was also exceptional.

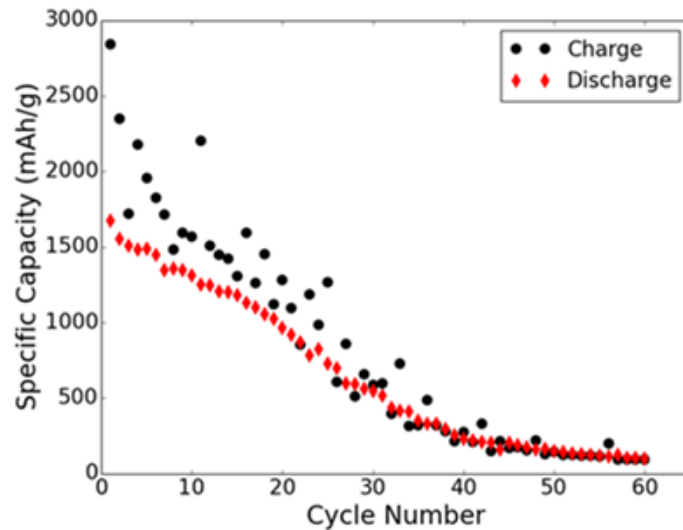
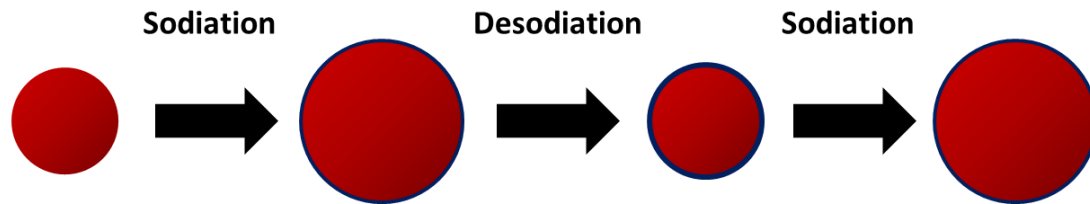


# Effect of different graphene content and chemical bonding in the P/graphene composite

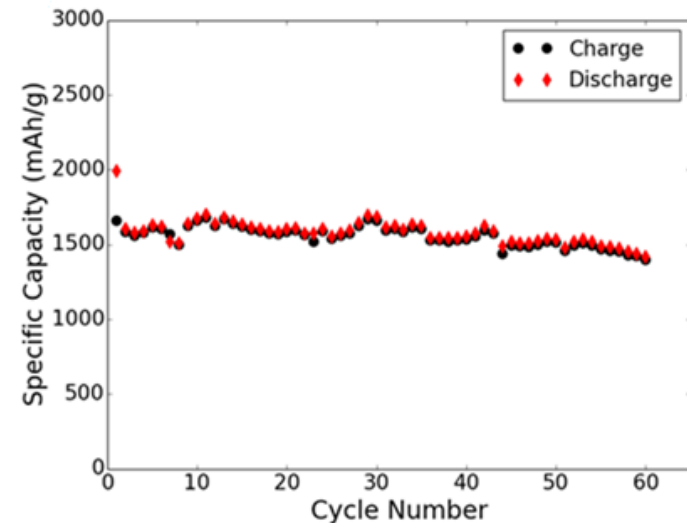


- (1) With the increasing Graphene content, the electrochemical performance are significantly improved.
- (2) Chemical bonding also plays an important role on the battery performance.

# Fluoroethylene carbonate (FEC) can help form a stable SEI layer to improve cycling stability

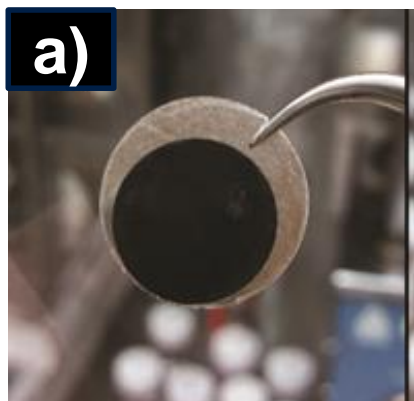
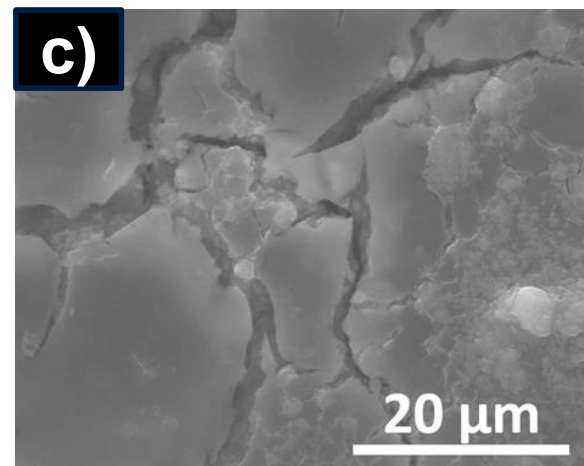


**FEC-Free**

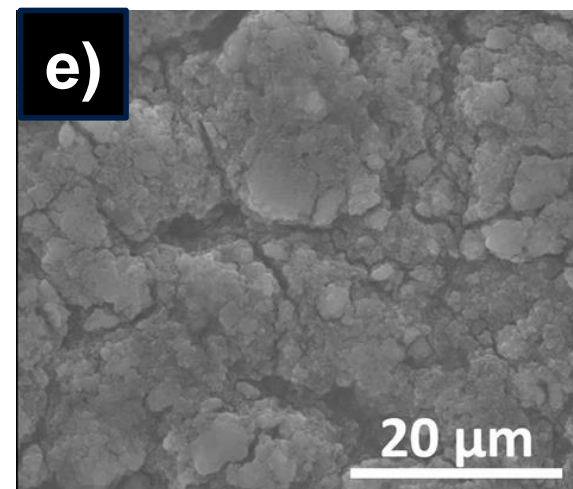


**FEC-Containing**

# Optimization of Electrolyte (1M NaClO<sub>4</sub> in EC/DEC+ 10wt% FEC)

**a)****Without FEC****b) After 20 cycles****Significant Na deposition Particles Un-observable****c)**

20 μm

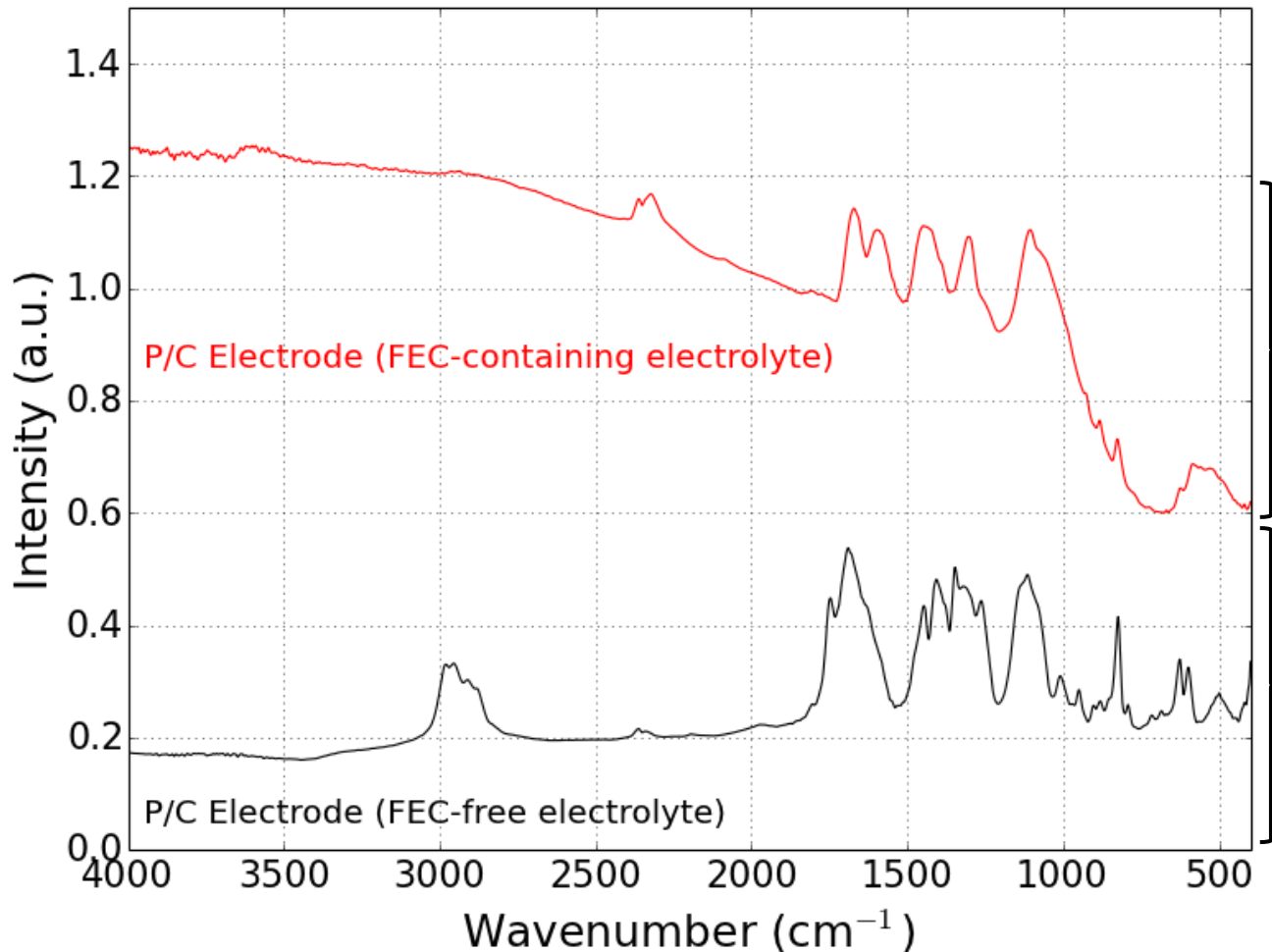
**d) After 20 cycles****No Na metal deposition****e)**

20 μm

**Particles Observable**



# Using FTIR, more robust SEI species were found to form with FEC.



Both have some  
ROCO<sub>2</sub>Na, RCO<sub>2</sub>Na

**FEC-containing  
has more:**

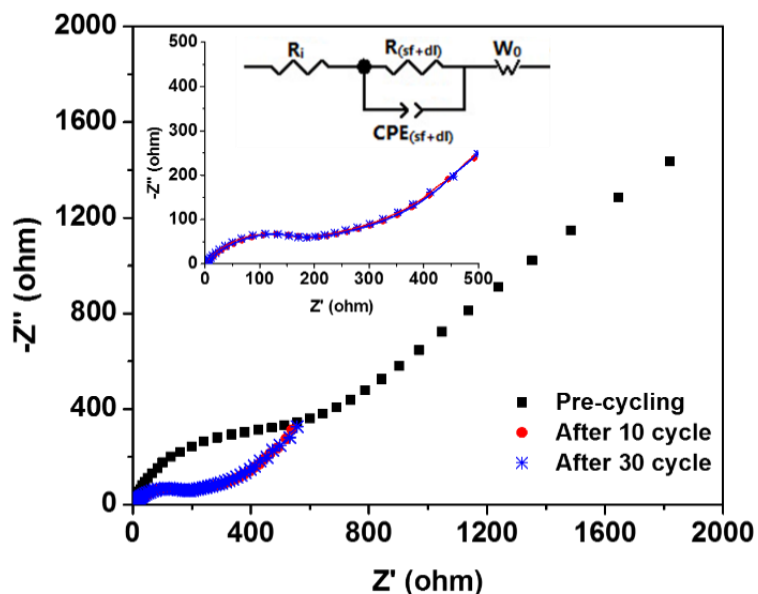
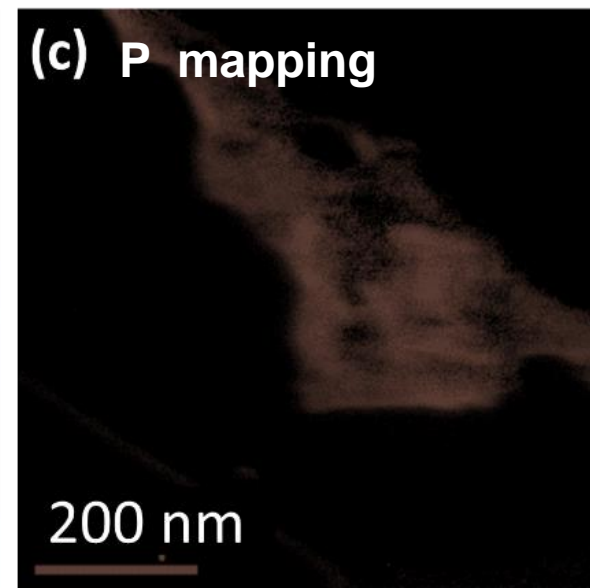
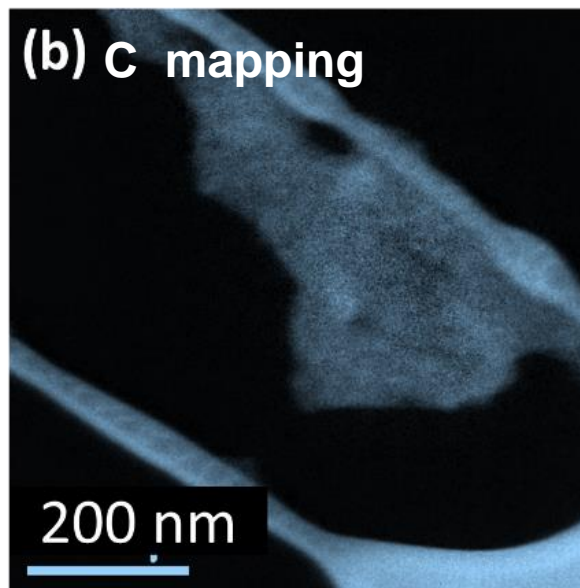
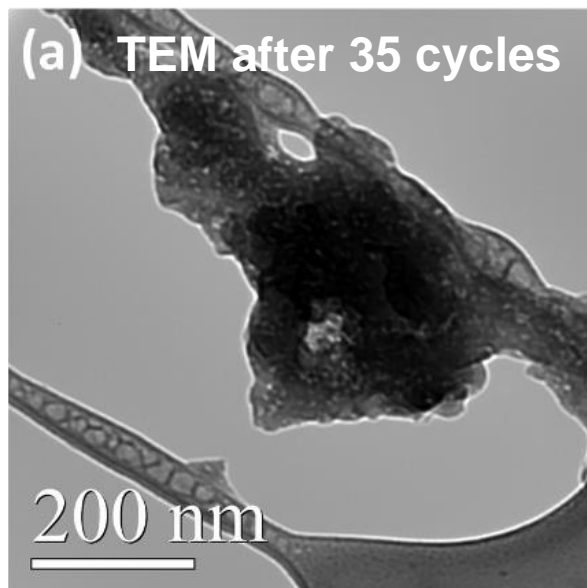
- RCO<sub>2</sub>Na
- Polycarbonates

**FEC-free has more:**

- RONA
- Monocarbonate
- ROCO<sub>2</sub>Na (possible)
- Ester or alkyl carbonate

**More robust species with FEC**

# Post-cycling characterization



(1) The almost overlapped EIS spectra indicate a good maintenance of conducting electrical contact and a relatively stable SEI.

(2) The graphene still has a uniform distribution in the composite and provides a good conducting matrix upon cycling.

# Summary

- 1) Graphene-nanosheet-wrapped phosphorus composite anode for sodium-ion batteries via a simple ball-milling approach with low-cost precursors of red phosphorous and graphene stacks
- 2) The featured structure and chemical bonding (P-O-C) between P and G play an important role on the electrochemical performance of P-Based anode for SIB.
- 3) More stable SEI formation using FEC



**Progress toward sodium-ion batteries  
For stationary energy storage**

# Acknowledgements

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