Zn-Air Batteries for Long-Duration Energy Storage

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Motivation and Objectives

- Efficient use of renewable sources necessitates to develop a low-cost and sustainable energy storage technology.
- The proposed novel Zn-air battery (ZAB) system has a strong market demand because of its easily maintained supply chain, low-cost (\$10 KW h), high-energy density $(1086 \text{ Wh kg}^{-1}).$
- Major short comes of Zn-air battery are severely limited operating life and low round trip energy efficiency.
- Need to overcome key challenges related to material development, processing, electrode and electrolyte design, and cell engineering that hinder the practical applicability of these novel Zn-air batteries.
- Scale up the synthesis of low-cost PGM free bifunctional electrocatalysts.
- Developing a cross-linking composite polymer membrane that enables to minimize the water loss and keeps the high conductivity.
- The goal is to develop a low cost, critical material free, multifunctional ZAB prototype that can be integrated into renewable energy technology platforms to enable long duration energy storage.

Synthesis of $Ni_{0.95}Fe_{0.05}(OH)_2$ and $Ni_4Fe_{1.5}Re_{0.6}O_x$ by coprecipitation and solution process **Solution C** N₂ PUMP Sodium Hydroxide in **Solution B** Sodium Hydroxide in Continuous Ammonia Solution **Solution A** Transition Metal Solution in Ni. Fe - 2Mi, Fe **XRD Difference** P-3m1 **Reactor Solution** Ammonia in DI Water pH Maintained □catalyzer synthesized using 2M NaOH as the base

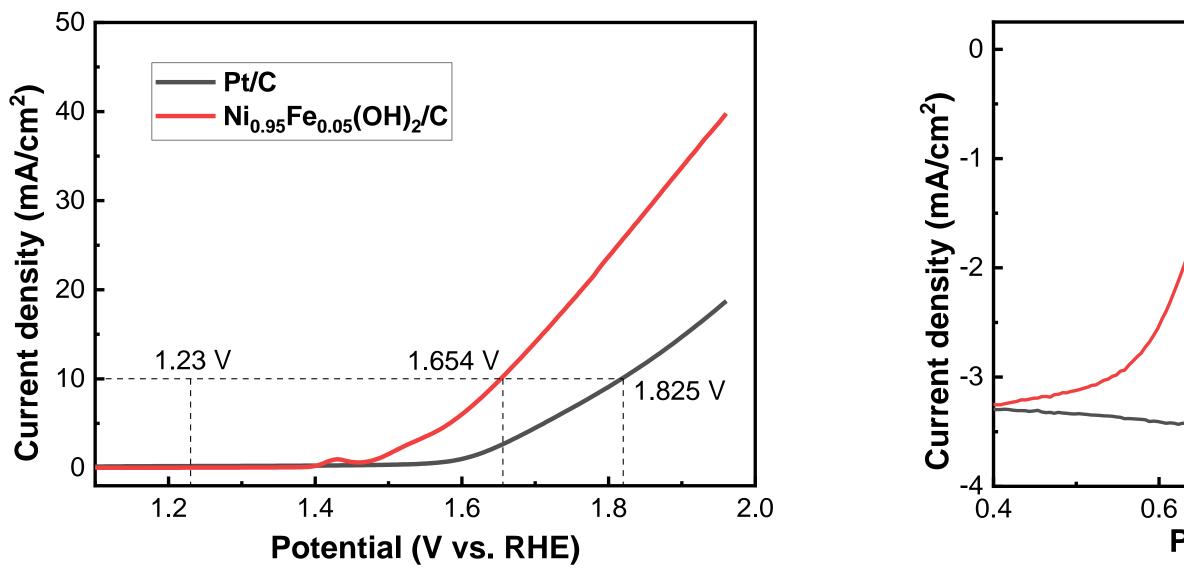
Synthesis of bifunction electrocatalysts

Zn-air battery cell

Zn foil Catalyst embedded **GPE** carbon paper

Electrochemical characterization of the Ni_{0.95}Fe_{0.05}(OH)₂ and Ni₄Fe_{1.5}Re_{0.6}O_x

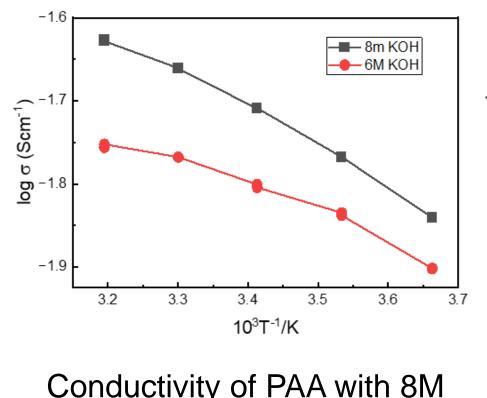
2θ (deg)



OER overpotential value for Ni_{0.95}Fe_{0.05}(OH)₂/C: 424 mV and Pt/C: 595 mV at 10 mA/cm², respectively. ORR performance were evaluated in O_2 saturated 0.1 M KOH, showing $E_{1/2}$ at 0.645 V and 0.852 V.

Conductivity and electrochemical stability of fabricated composite gel polymer electrolyte (GPE)

 $\square N_2$ gas pumped to avoid iron oxidation – Improving

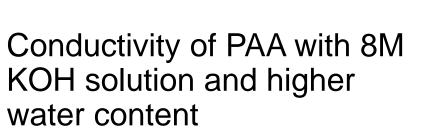


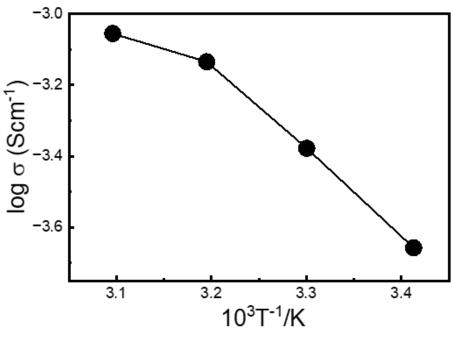
 $Zn(OH)_4^{2-}$ + $H_2O + 2OH^{-}$

 Zn^2

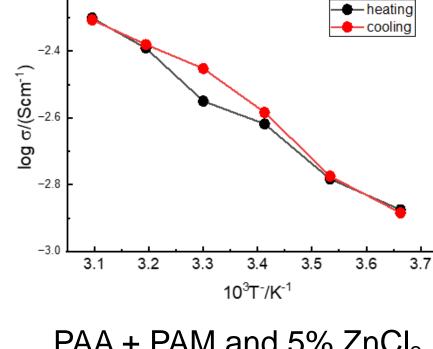
40H

 $Zn^{2+} - Zn(OH)^{2-}$

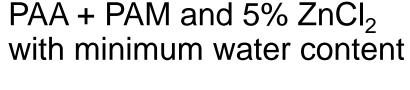


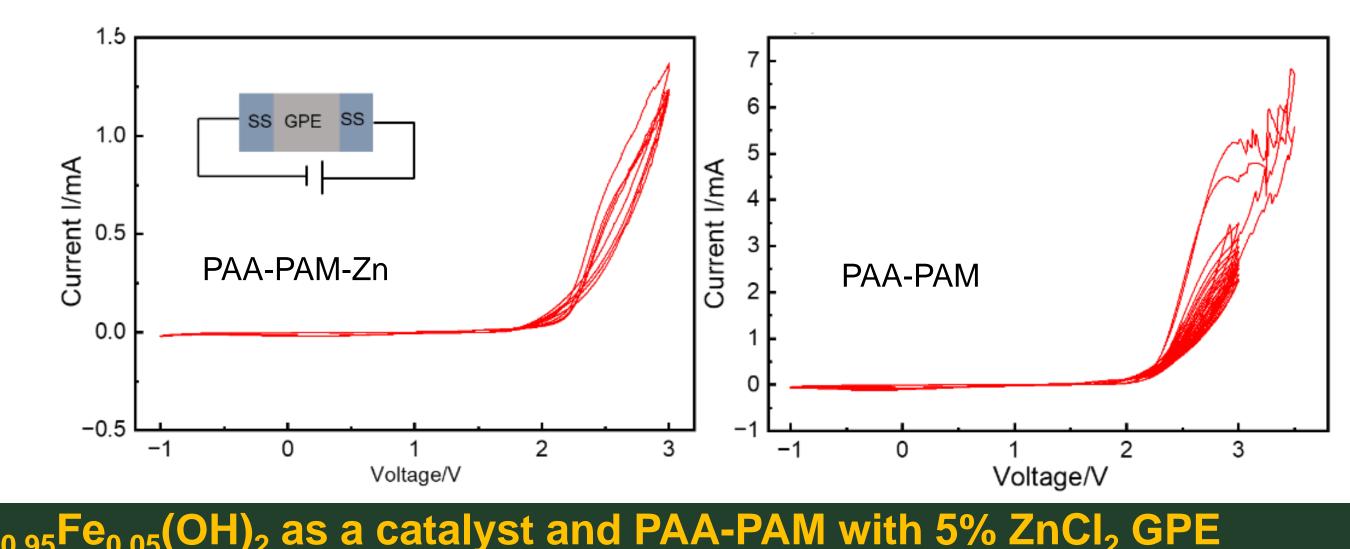


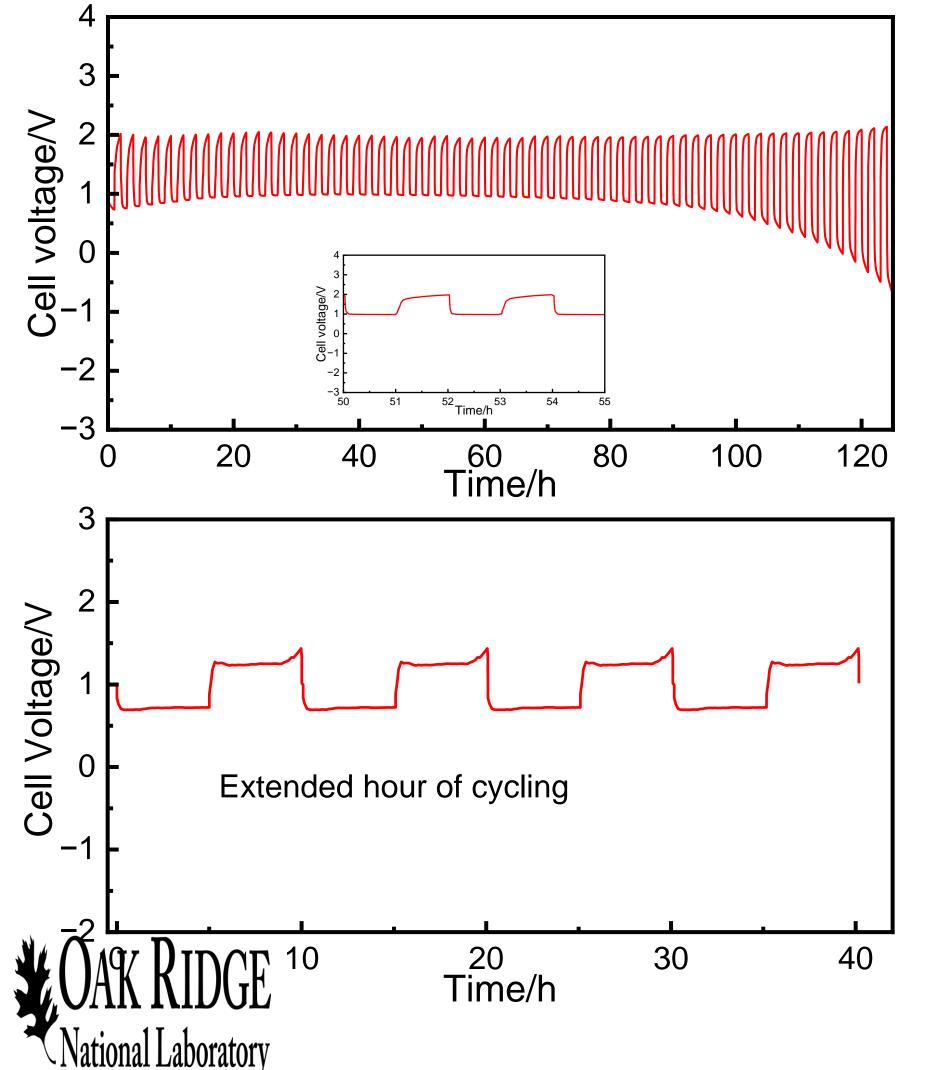
Conductivity of PAA with 8M KOH solution minimum water content Electrochemical cycling performance of Zn-air battery using Ni₀

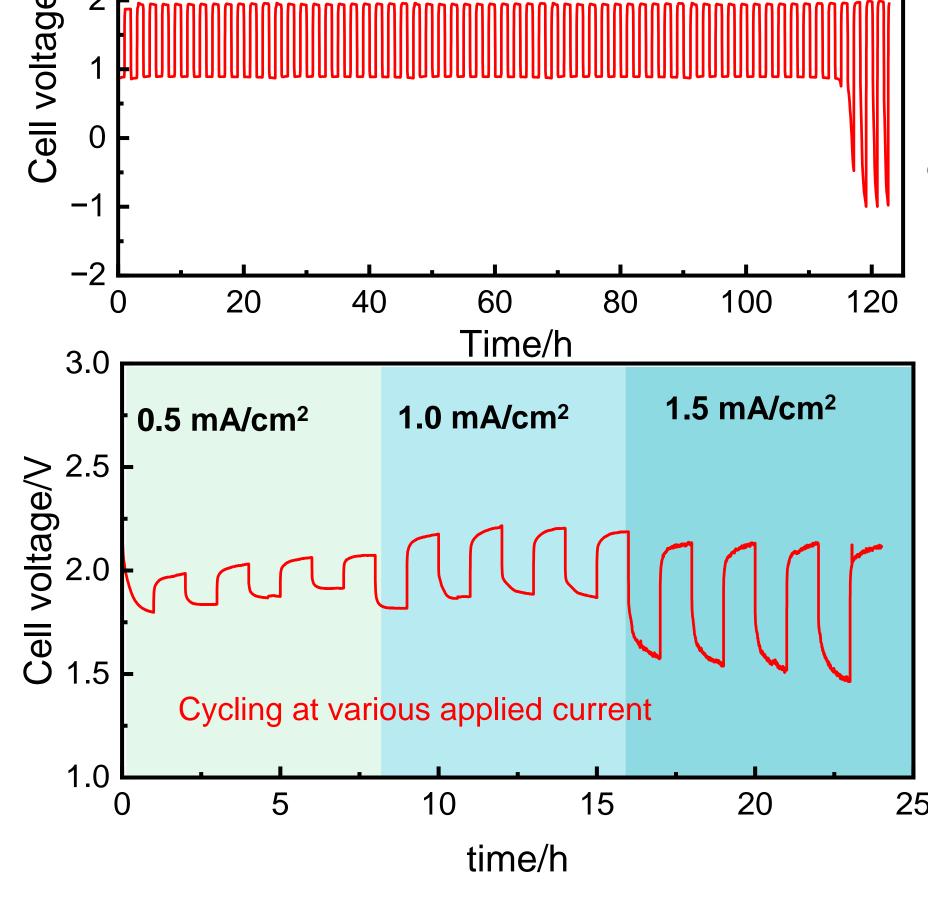


morphologies









voltage/V 100 200 500 300 400 Time/h

- Pt/C

Potential (V vs. RHE)

 $Ni_{0.95}Fe_{0.05}(OH)_2/C$

1.0

- ***** Both $Ni_{0.95}Fe_{0.05}(OH)_2$ and $Ni_4Fe_{1.5}Re_{0.6}O_x$ exhibit similar catalytic activity for OER and ORR.
- Cross linking composite gel polymer electrolyte shows better performance due to higher water retention ability.
- Better cycling performance is obtained for composite GPE and Ni_{0.95}Fe_{0.05}(OH)₂ catalyst.
- Two voltage steps slope is appeared due to the new interphase formation.